# Reduced Precious Metal Catalysts for Methane and NOx Emission Control of Natural Gas Vehicles

Mike Harold (PI), Lars Grabow (University of Houston)
Steve Golden (CDTi)

Bill Epling (University of Virginia)

Bill Partridge, Josh Pihl (Oak Ridge National Laboratory)

June 12, 2019









**ACE128** 

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#### **Overview**

#### **TIMELINE**

Start: May 1, 2018End: April 30, 2021

■ 30% complete

#### **BUDGET**

Total project funding

■ DOE: \$2,000,000

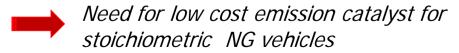
■ UH & partners: \$525,380

Funding received

■ FY19: \$660,176 (by 4/1/2019)

#### BARRIERS/TARGETS

- Replace/supplement diesel with domestic natural gas (NG) for transportation
- Methane is greenhouse gas (25x CO<sub>2</sub>); CH<sub>4</sub>
   GHG emissions above the 30 mg/mi light- duty vehicle cap count against fuel economy
- U.S. EPA mandates tailpipe methane emissions at 0.1 g/bhp-h for heavy-duty vehicles (95% reduction) & NOx emissions at 0.2 g/bhp-h
- State-of-art three-way catalyst (TWC) ineffective for methane oxidation at < 400 °C</p>



#### **PARTNERS**

- U. Houston (lead)
- CDTi Advanced Materials, Inc.
- University of Virginia
- Oak Ridge National Lab

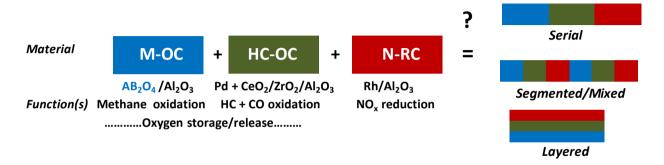




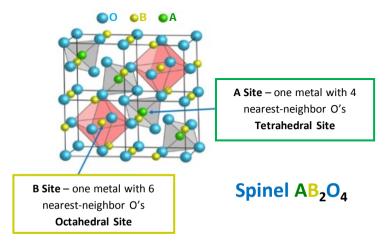


### **Project Premise and Hypothesis**

Develop the: FWC = Four Way Catalytic Converter to enable reduced emissions of  $CH_4$  in addition to CO, NOx & NMHCs from CNG-fueled vehicles

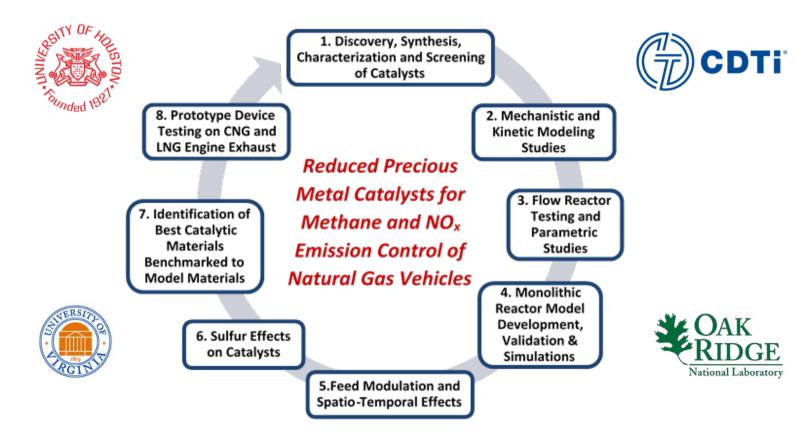


Spinel in combination with low levels of precious metals are cost-effective solution for coupled methane, CO and NO<sub>x</sub> conversion in stoichiometric natural gas vehicle exhaust.





### **Project Approach and Collaborations**



Comprehensive program spanning discovery, development, evaluation, and technology transfer will help to bring down cost barriers and accelerate the deployment of NG vehicles in the medium- and heavy-duty sectors



### **BP1 Milestones**

#### Milestone

#### **Description**

#### **Status**

**Material Discovery** Complete

Identify at least one Spinel material using descriptor-based density functional theory (DFT).

Several Spinel materials have been evaluated and rank-ordered based on methane oxidation activity. DFT screening efforts are underway to compare with measurements

Reference and Baseline material performance testing Complete

Document flow reactor performance testing of Reference and Baseline FWC materials in terms of methane. CO, and NO<sub>v</sub> conversion.

Reference and Baseline catalysts have been evaluated for both steady state and modulation conditions.

Kinetic model tuning Complete

Develop, tune and validate monolith reactor model containing the global kinetics of the Baseline FWC material.

Experiments are underway to compare activities of different functions of the Baseline material comprising Spinel and precious group metal (PGM). Kinetics evaluations are underway as first step towards kinetic model development.

Identification of candidate material (Go/No Go)

off below 300 °C and has less than 30 g PGM/ft<sup>3</sup> at 50k h<sup>-1</sup> space velocity (following USDRIVE specified protocol, if available) identified.

At least one FWC material that lights Several FWC materials comprising PGM toplayer and Spinel base-layer have demonstrated light-off at ~300 °C at 90 h<sup>-1</sup> space velocity using realistic feed with PGM loading of 30 g/ft<sup>3</sup>.



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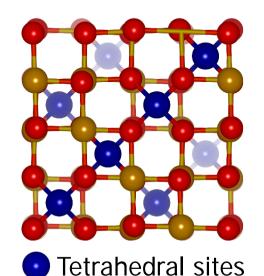
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# DFT Descriptors for CH<sub>4</sub> Conversion



Octahedral sites

#### Oxygen vacancy formation energy

- indicates lattice oxygen release/uptake

#### H-binding energy

- indicates C-H bond activation (a key step for hydrocarbon oxidation)

Both descriptors closely related to structure and composition of the Spinels, and can be used for performance screening

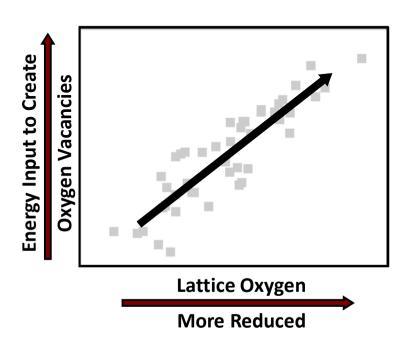
#### Structure - composition space of spinel oxides $(A_{1+x}B_{2-x}O_4)$

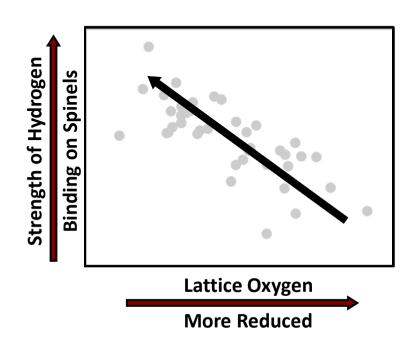
Structural Phases: Normal; Semi-Inverse; Inverse

Compositions:
 A, B = {Mn, Fe, Co, Ni}



### **Descriptor-based Spinel Screening**





#### Oxygen vacancy formation energy

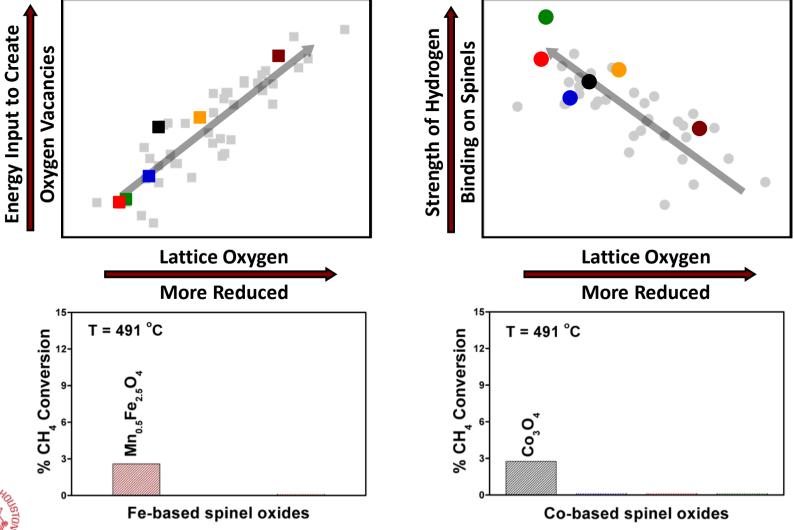
Higher energy required to create oxygen vacancies in materials with highly reduced lattice oxygen

#### H-binding energy

Stronger H-binding observed over materials with less reduced lattice oxygen



### **CH**<sub>4</sub> Conversion Performance





### **Summary: Descriptor DFT**

- DFT reliable for screening candidate Spinels
- Experimental trends along with DFT study portend need for:
  - Lower oxygen vacancy formation energies (marks low temperature vacancy creation)
  - Stronger H-binding (crucial for C-H bond activation – primary step for HC oxidation)



### **BP1 Milestones**

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### **Kinetics & Flow Reactor Studies**

Baseline Catalyst: 30

**30** g PGM/ft<sup>3</sup>

100 g Spinel/L

**"30/100"** 

Steady

 $\lambda = 0.995$ 

**Feeds** 

95% Pt, 5% Pd

 $Mn_{0.5}Fe_{2.5}O_4$ 

"Spinel": 25% Spinel

75% γ-Al<sub>2</sub>O<sub>3</sub>

 $<\lambda> = 0.995$ (0.981-1.009 @ 0.33 Hz)

 $\lambda = \frac{0.5 \times [CO + 2^*O_2 + NO + H_2O + 2^*CO_2]}{[CO + CH_4 + CO_2] + 0.25 \times [2^*H_2 + 4^*CH_4 + 2^*H_2O]}$ 

**CH<sub>4</sub>** 1500 ppm

1500 ppm

**Modulated** 

**CO** 8000 ppm

8000 ppm

 $H_2$ 

2000 ppm

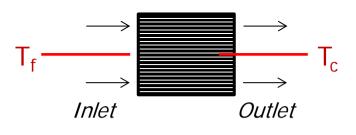
2000 ppm

NO

1000 ppm

1000 ppm

**O<sub>2</sub>** 6700 ppm 4500 ppm (1.5 s) 8900 ppm (1.5 s)



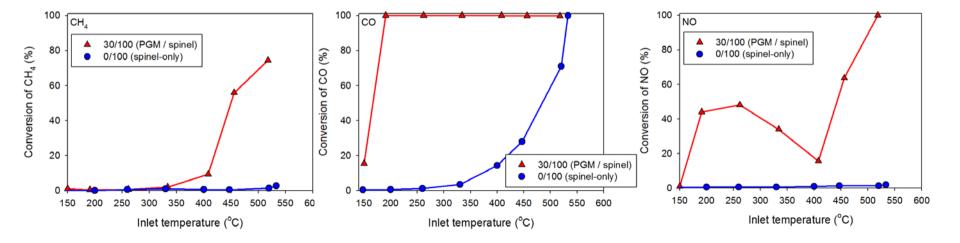
Monolith Catalyst

 $(1 \times 0.84 \text{ in.})$ 

Background: 10% H<sub>2</sub>O, 10% CO<sub>2</sub>, bal. N<sub>2</sub>

### Baseline Catalyst: Impact of Ingredients

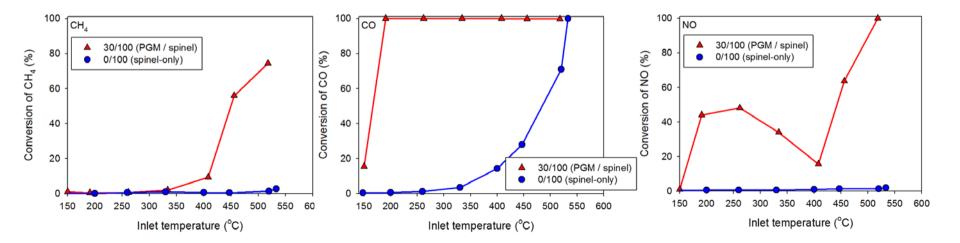
Baseline (CDTi) X/100 X = 0, 30 g PGM/ft<sup>3</sup>, 100 g Spinel/L GHSV = 40k hr<sup>-1</sup> Steady Feed  $\lambda$  = 0.995





### Baseline Catalyst: Impact of Ingredients

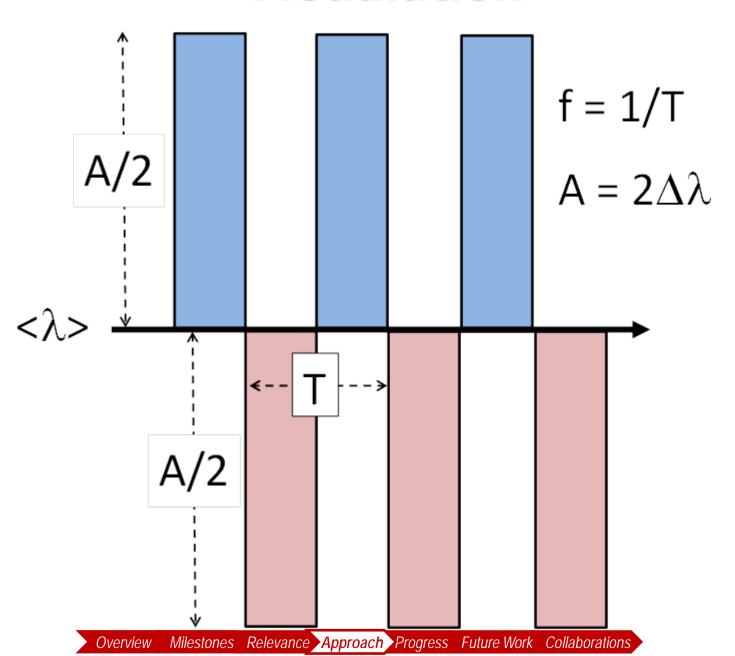
Baseline (CDTi) X/100X = 0, 30 g PGM/ft<sup>3</sup>, 100 g Spinel/L GHSV = 40k hr<sup>-1</sup> Steady Feed  $\lambda$  = 0.995



• For steady feed, PGM is active component;  $Mn_{0.5}Fe_{2.5}O_4$  Spinel is not



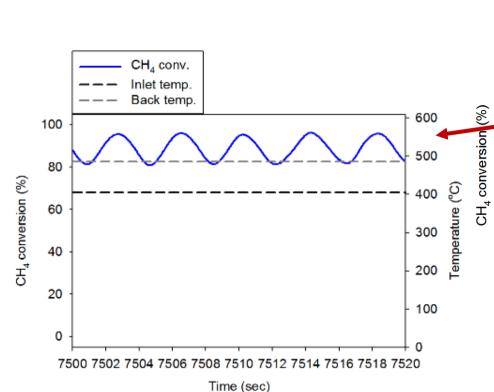
### **Modulation**



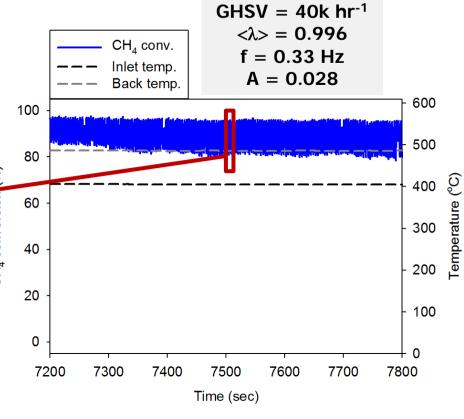


### **Modulation Impact**

Baseline (CDTi) 30/100 30 g PGM/ft<sup>3</sup>, 100 g Spinel/L







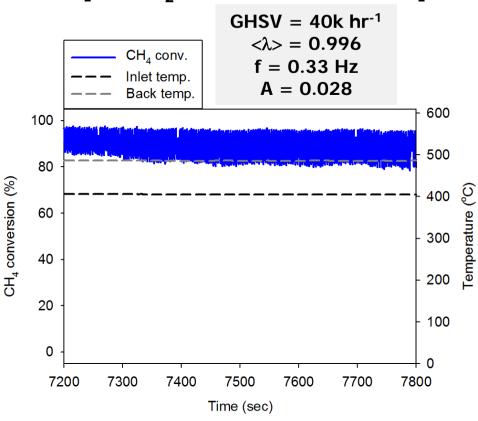


### **Modulation Impact**

#### [w/o O<sub>2</sub> modulation at 405 °C]

#### $GHSV = 40k hr^{-1}$ $<\lambda> = 0.996$ CH, conv. f = n/aInlet temp. A = 0Back temp. 600 100 500 80 CH<sub>4</sub> conversion (%) Temperature (°C) 400 60 300 40 200 20 100 0 0 5500 5700 5800 5900 5400 5600 6000 Time (sec)

#### [with O<sub>2</sub> modulation at 405 °C]

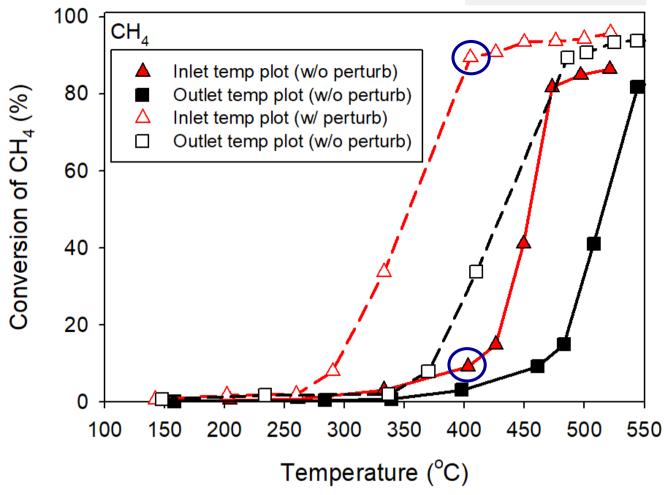


• Modulation significantly enhances PGM + Mn<sub>0.5</sub>Fe<sub>2.5</sub>O<sub>4</sub> Spinel activity



### **Baseline Catalyst Performance**

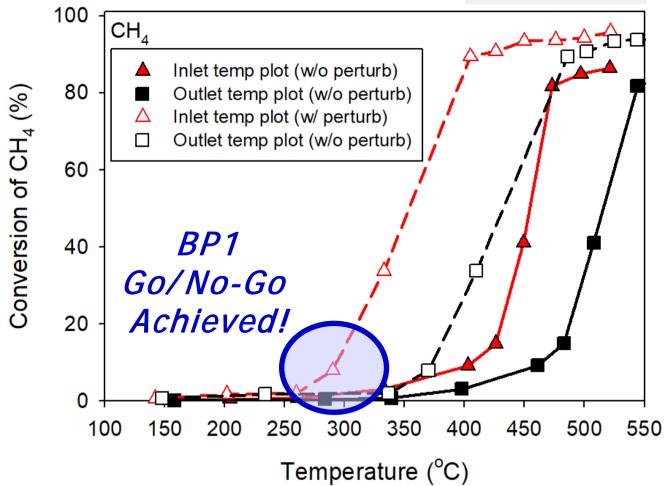
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### **Baseline Catalyst Performance**

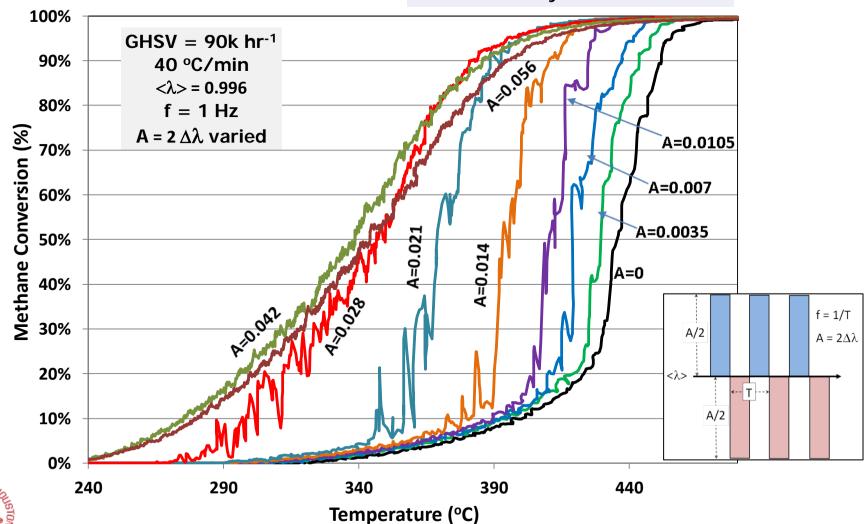
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### Impact of Modulation Amplitude

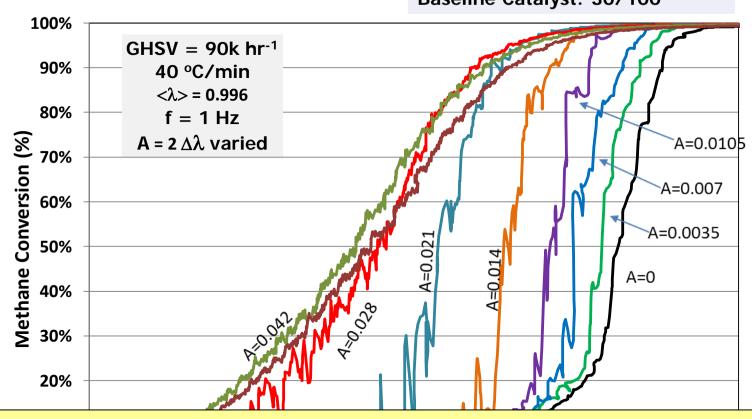




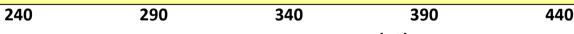


### Impact of Modulation Amplitude

**Modulation Amplitude Variation** CH4 CONVERSION Baseline Catalyst: 30/100



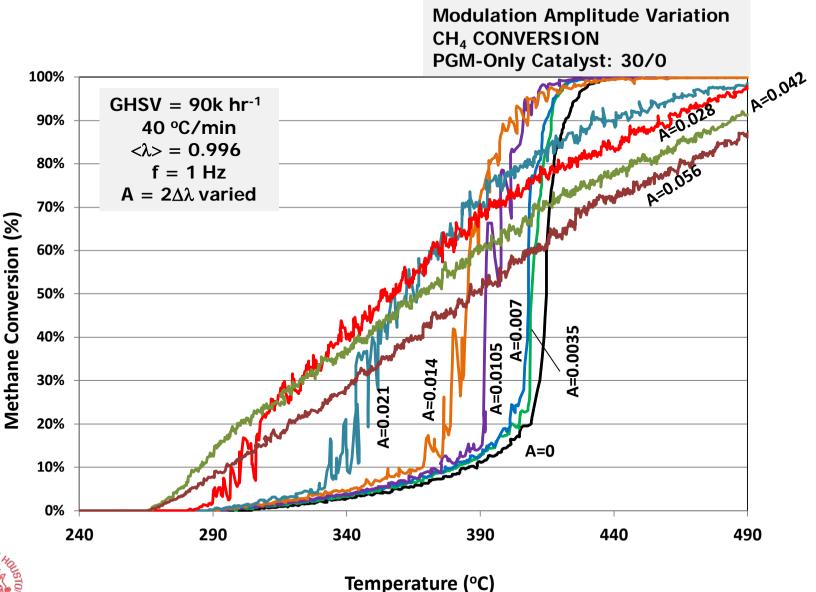
- PGM + Spinel shows impressive enhancement with modulation
- Modulation amplitude tuned to maximize enhancement





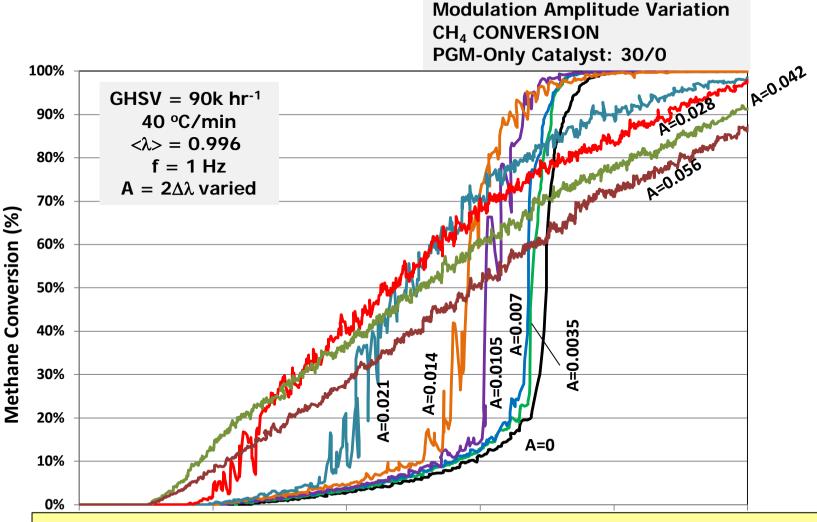


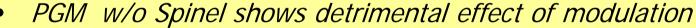
### Impact of Spinel





### Impact of Spinel





Spinel is essential ingredient to achieve low light temperature



### Spinel + PGM Coupling: Layered vs. Mixed Washcoat

PGM (30 g/ft<sup>3</sup>) + Spinel Dual-Layer

30/100 PGM/Spinel

PGM/Alumina

Spinel

Cordierite

**Top layer** (100 g/L; 1.63 g/in<sup>3</sup>)

Bottom layer (100 g/L; 1.63 g/in<sup>3</sup>)



PGM (30 g/ft<sup>3</sup>) + Spinel Mixed Layer

PGM/Alumina + Spinel

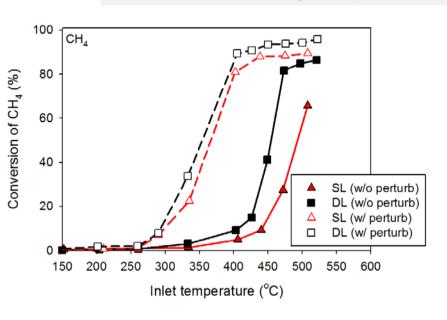
Cordierite

**Total:** 200 g/L (3.26 g/in<sup>3</sup>)

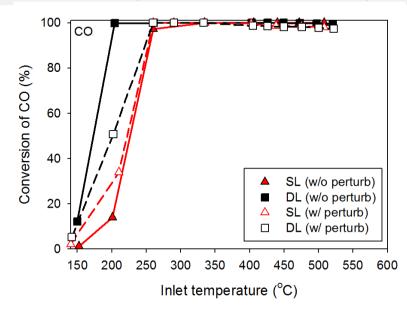


### Impact of PGM & Spinel Proximity

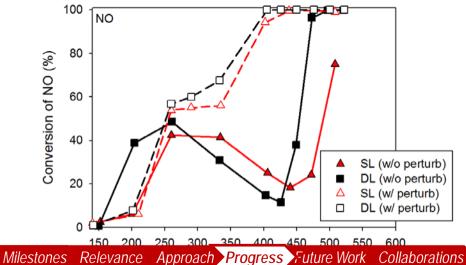
Comparison: Single-layer (SL) vs. Double-layer (DL) 30/100 Catalysts



Overview



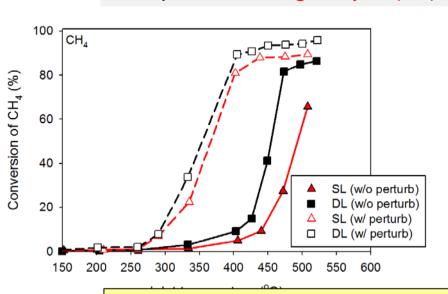
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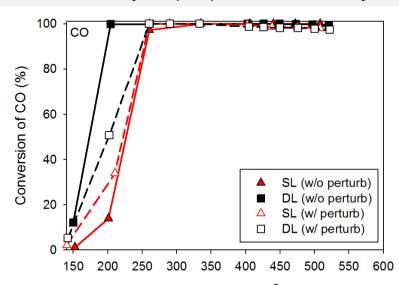


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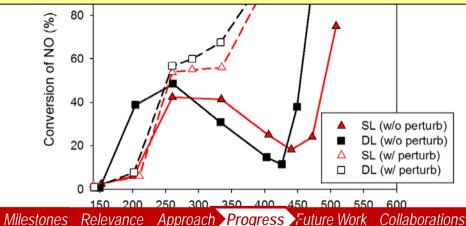


Overview



- Dual-layer and Single-layer PGM + Spinel show similar results
- Close proximity of PGM & Spinel not essential

GHSV =  $40 \text{ hr}^{-1}$   $<\lambda>=0.995$  f=0, 0.33 HzA=0, 0.028





# Conversion Enhancement: Observations

- PGM (at 30 g/ft³) is much more active than Spinel (at ~100 g/L)
- PGM activity is enhanced by Spinel under modulation
- Modulation can be tuned to achieve maximum enhancement
- Mechanistic origin of enhancement related to strong dependence of methane activity on O₂ concentration:

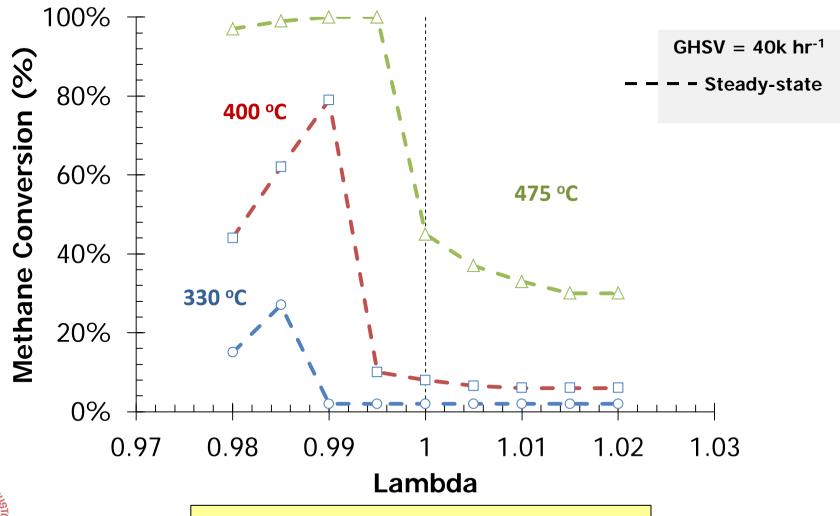
Key may be Spinel oxygen storage & release keeps PGM in the more active mixed metallic-oxide state



### Lambda Sweep: Methane Conversion

PGM-only Catalyst (30 g/ft<sup>3</sup>; 60 g/L PGM)

 $L_I = 0 g/L$ 



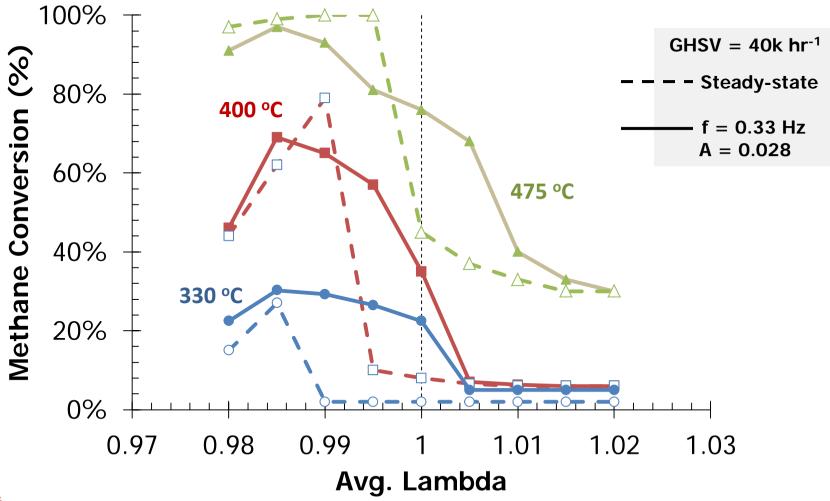


Strong  $O_2$  inhibition as  $\lambda \rightarrow 1$ 

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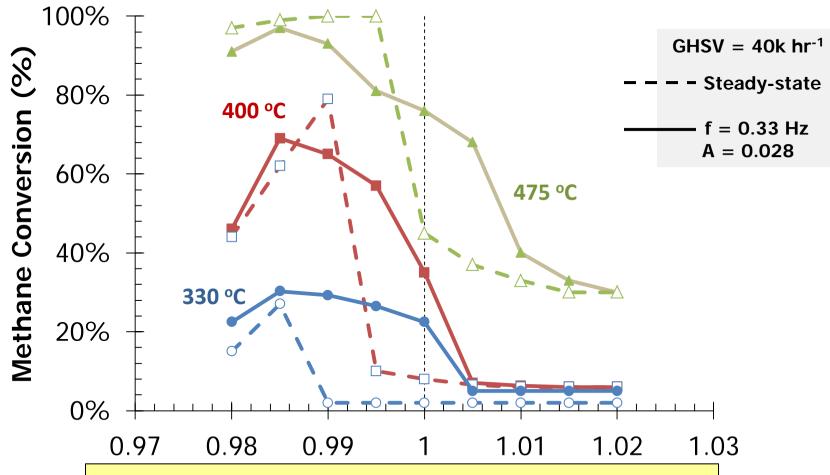




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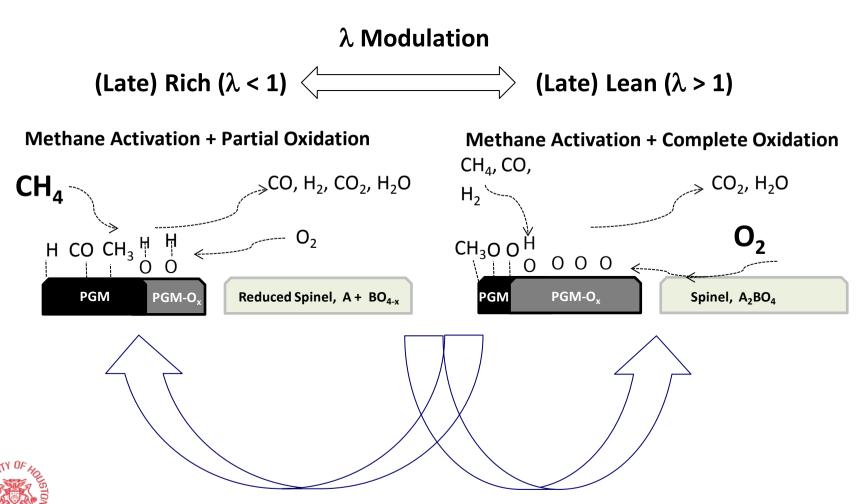
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- Strong  $O_2$  inhibition as  $\lambda \rightarrow 1$
- Modulation enhancement for PGM only catalyst

# Conversion Enhancement: Proposed Mechanism



### Remaining Challenges & Barriers

- Lower light-off below 300 °C with PGM < 30 g/ft<sup>3</sup>
- Understand mechanism for Spinel enhancement
  - Direct and/or Indirect (methane oxidation) (oxygen storage/release)
- Increase Spinel methane oxidation activity& dynamic oxygen storage capacity
- Understand spatial trends during modulation
- Develop predictive model to guide improvements
- Address sulfur tolerance and mitigation



### **Proposed Future Work**

**Task 2.1** – Discover new Spinel materials using computational screening.

Descriptor-based DFT will be accelerated in order for the computations to drive synthesis efforts. Rank-ordering of Spinel compositions is goal.

**Task 2.2** – Synthesize, characterize and screen candidate Spinel materials.

Effort will continue based on experimental screening of Spinel powders with guidance from the computational efforts.

Task 2.3 – Develop mechanistic kinetic models of FWC materials.

Kinetic modeling will continue for the Baseline and Reference materials, then move to the more active Spinel materials. Will involve conventional kinetics measurements, oxygen uptake and release kinetics, and Temporal Analysis of Products (TAP) experiments.



### **Proposed Future Work**

**Task 2.4** – Carry out flow reactor parametric studies of FWC materials.

■ Focus on new FWC materials with Baseline and Reference materials complete.

**Task 2.5** – Carry out spatially-resolved flow reactor studies of Baseline FWC materials.

Spatially-resolved measurements of the Baseline and Reference material will be conducted to provide data for modeling efforts.

**Task 2.6** – Perform stoichiometry modulation studies of FWC materials.

Attention will be placed on new materials.



### **Proposed Future Work**

**Task 2.7** – Evaluate sulfur exposure on Baseline and new FWC materials.

Comprehensive study of sulfur exposure will be conducted during this period to identify sulfur-tolerant materials.

**Task 2.8** – Carry out modeling of monolith reactors containing Baseline Spinel materials.

This is a key task for BP2 to set the stage for model-directed optimization later in the project.

**Task 2.9** – Identify the best catalytic materials.

This task spans all other tasks with convergence to the most active catalyst formulation the end goal.



### Summary

#### ■ Relevance

■ Enabling emergence of natural gas vehicles by removing emissions hurdle

#### Approach

■ From molecular-level discovery & mechanism to development & demonstration

#### ■ Technical Accomplishments & Progress

- Good progress on all fronts; BP1 milestones achieved
- PGM + Spinel initial results encouraging; even better materials targeted

#### Collaborations & Coordination

■ Cooperation: universities (UH+UVA), national lab (ORNL), industry (CDTi)

#### Proposed Future Research

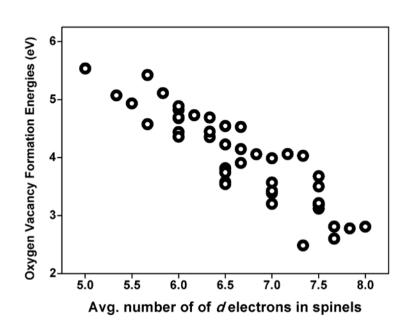
■ Converge on next-gen catalysts, integration, modeling, & optimization

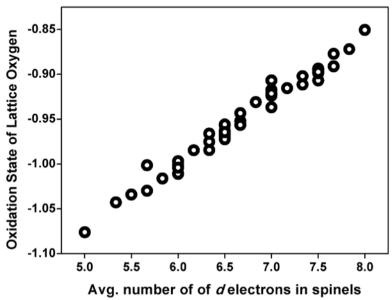


### **Technical Backup Slides**



### d Electrons – a Key Intrinsic Property

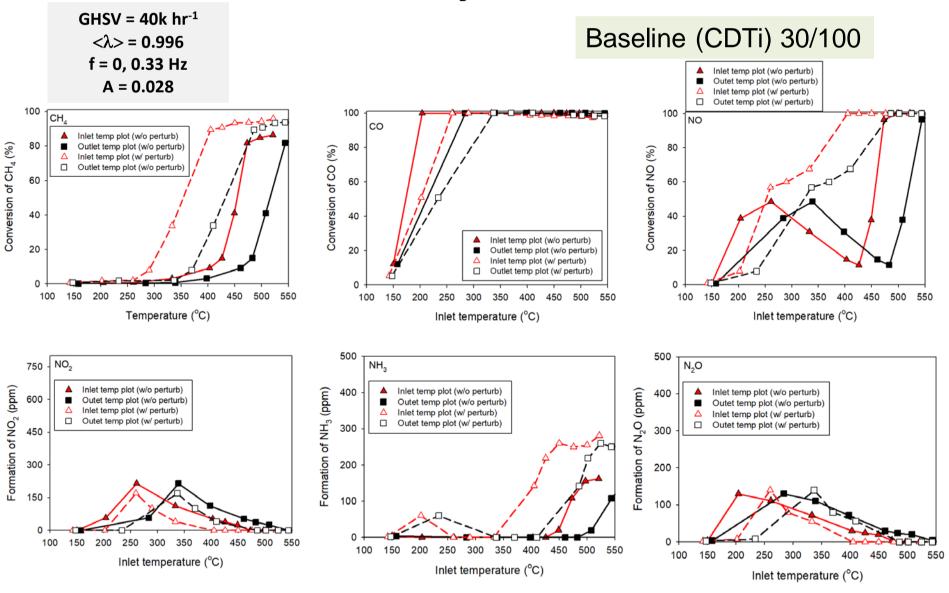




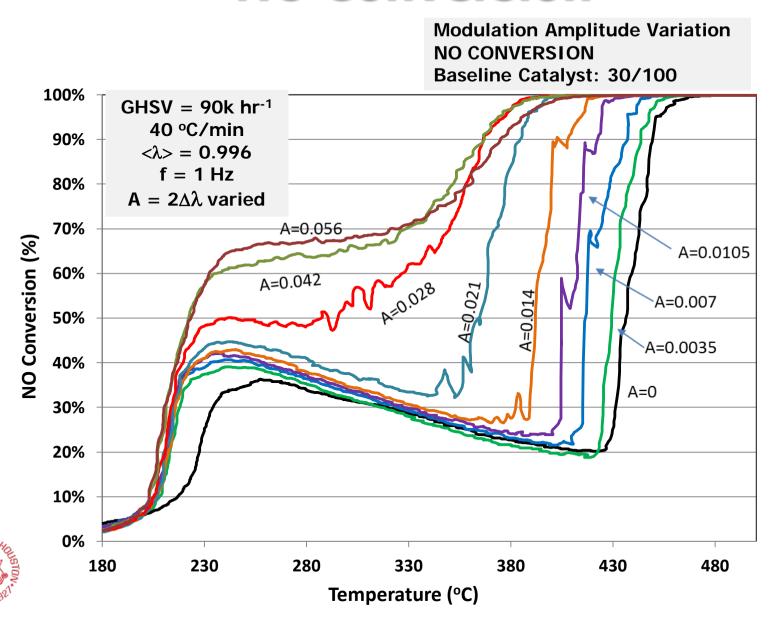
- Number of d electrons shows direct correlation to:
  - oxygen vacancy formation energy and
  - oxidation state of the lattice oxygens
- This allows for prediction of desired materials without rigorous DFT calculations



### **Baseline Catalyst Performance**

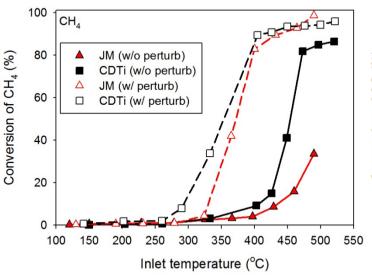


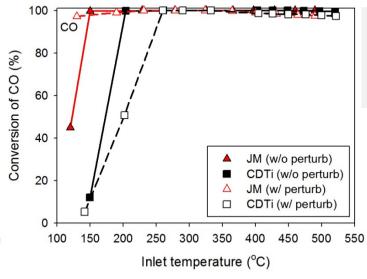
## Impact of Modulation: NO Conversion



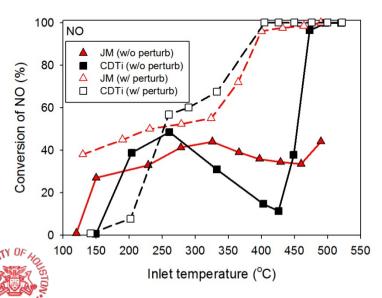
### **Catalyst Benchmarking**

Comparison: Reference (JM) vs. Baseline (CDTi) 30/100



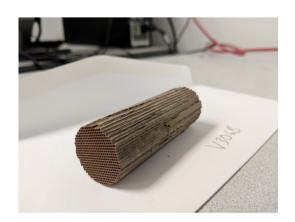


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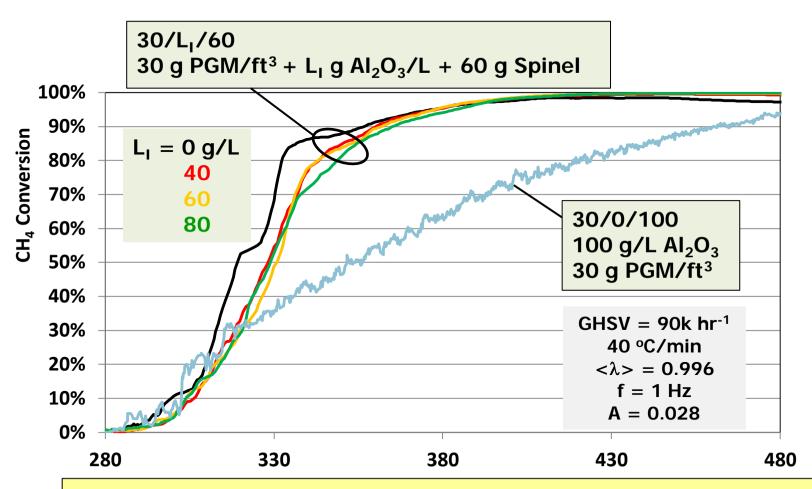






PGM/Spinel/Al<sub>2</sub>O<sub>3</sub>(CDTi) 30 g {Pt(95%)+Pd(5%)}/ft<sup>3</sup>

### Impact of Intermediate Layer



- Intermediate layer does not have significant impact
- Spinel required to give enhanced activity